

A new tryptophan derived alkaloid from *Evodia rutaecarpa* (Juss.) Benth. et Hook

B. Danieli, G. Lesma and G. Palmisano

Istituto di Chimica Organica, Centro CNR di Studio per le Sostanze Organiche Naturali, Università degli Studi di Milano, via Saldini 50, I-20133 Milano (Italy), 7 July 1978

Summary. The first tryptophan-derived indolopyridoquinazoline alkaloid has been isolated from the fruits of *Evodia rutaecarpa* (Jussieu) Benth. et Hook. Its structure is deduced on the basis of spectral data and enantioselective synthesis. The compound has been assigned the absolute configuration (7*S*, 13*bS*)-7-carboxy-8, 13, 13*b*, 14-tetrahydro-14-methylindolo[2', 3': 3, 4]pyrido[2, 1-*b*]quinazolin-5(7*H*)-one.

Although most indole alkaloids are formally derived from tryptamine, there have recently been discovered a rapidly expanding class of non-isoprenoid and monoterpene alkaloids, which retain the carboxylic group of their tryptophan precursor¹.

In continuation of our earlier studies² on the quinazoline alkaloids, we report on the isolation and structure determination of the first indolopyridoquinazoline alkaloid ascribable to this class of tryptophan-derived compounds. After isolating rutaecarpine and evodiamine by neutral alumina chromatography of the methanolic extract of the fruits of *E. rutaecarpa* (Juss.) Benth. et Hook, methylation (CH₃N₂, MeOH-CHCl₃, r.t., 24 h) of the acidic chromatographic residue and purification by repeated preparative TLC on silica gel (benzene:diethyl ether, 1:1, R_f 0.44) gave a small amount (8 mg from 3.5 kg of dried fruits) of an amorphous colourless compound C₂₁H₁₉N₃O₃ [by HRMS, M⁺ m/e 361.1417 (M⁺ calcd, m/e 361.1421)], [α]_D²⁰ + 441° (CHCl₃, c 1), CD (MeOH) λ_{max} 227 (Δε + 5.58), 257 (+ 11.89), 284 (+ 2.05), 292 (+ 3.10), 315 (+ 3.20), to which the structure **1b** was proposed on the basis of the following data.

The UV-spectrum (MeCN) (λ_{max} 268, 281, 289 and 330 nm) was identical to that of evodiamine³, pointing out the presence of an indolic chromophore and of an anthranilic one, both confirmed by ¹H-NMR signals (100 MHz, CDCl₃), 7 aromatic protons between δ 7.10–7.70, 1 aromatic proton as dd (J₁ 8.0 Hz, J₂ 1.5 Hz) at δ 8.16 and NH at δ 8.41. Besides a tertiary amide band at 1665 cm⁻¹ and a NH-band at 3475 cm⁻¹, the IR-spectrum (CHCl₃) showed a carbomethoxy group (1745 cm⁻¹), also confirmed by a singlet at δ 3.59 in the NMR-spectrum. The mass spectrum (EI, 70 eV, 100°) of **1b** supported strongly an evodiamine-type structure, showing peaks at m/e 134 (a) arising from N-methylantranilic moiety, while fragments at m/e 302 (M⁺ - CO₂Me), 300 (M⁺ - CO₂Me - 2H), 229 (b), 201 (c) were reminiscent of a tetrahydro-β-carboline substituted at α-position to N_b by a carbomethoxy group⁴. Furthermore, the NMR-spectrum showed a singlet for a N-CH₃ group at δ 2.46, a broad singlet (W_{1/2}, 3 Hz) at δ 6.22 for C-13*bH* and a typical ABX pattern at δ 3.22 (J_{AB} 16.0 Hz, J_{AX} 6.0 Hz), 3.55 (J_{AB} 16.0 Hz, J_{BX} 1.5 Hz), 5.76 (J_{AX} 6.0 Hz, J_{BX}

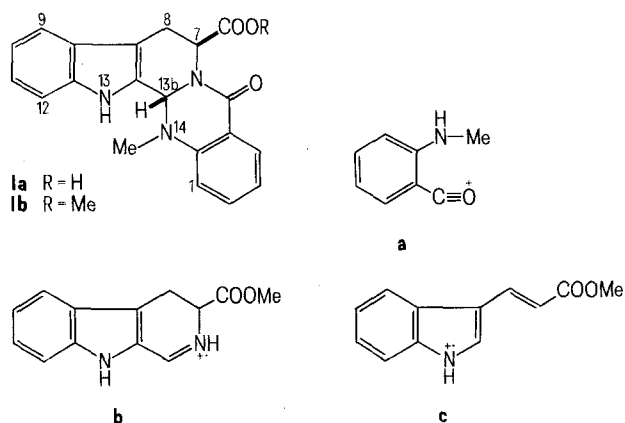
1.5 Hz) for C-8*Ha*, C-8*Hβ* and C-7*H*. Both C-8*Ha* and C-8*Hβ* signals exhibited a similar additional homobenzylic coupling (J ~ 1.5 Hz) with C-13*bH*, as confirmed by double resonance experiments.

The shielding and coupling constant values for the ABX system signals were accounted for only by assuming an axially oriented CO₂Me group and an equatorially oriented C-7*H* deshielded by the coplanar amidic carbonyl neighbouring group.

The carbomethoxy group localization at the C-7 and the absolute configuration of this centre were provided by synthesizing **1b** from (3*S*)-3-carbomethoxy-4,9-dihydro-3*H*-pyrido[3,4-*b*]indole hydrochloride and N-methylantranilic acid in THF containing 1 equiv. of pyridine in the presence of PPh₃-CBr₄⁵. A priori, one might have expected 2 C-13*b* epimeric compounds to obtain but, interestingly enough, only the 'natural' stereoisomer was formed (25% yield) under these mild conditions.

The observed stereospecificity could be explained on the basis of the steric and stereoelectronic requirements of the transition state(s). The carbomethoxy group of the tetrahydro-β-carboline from the β-equatorial to the β-axial orientation by the severe nonbonded interactions with the amide carbonyl group and, by analogy to the addition stereochemistry to olefinic type double bond, the NHCH₃ attack took place on the face opposite to the closer carbomethoxy group. **1b** was recovered unchanged after basic (MeONa, MeOH, r.t., 5 h) and acid-catalyzed (AcOH, reflux, 6 h) equilibrations suggesting that all the steric dipolar interactions and orbital overlap requirements were favorable only in the C-13*b*-β*H* epimer. Hence the structure **1b** of (7*S*, 13*bS*)-7-carbomethoxy-8, 13, 13*b*, 14-tetrahydro-14-methylindolo[2', 3': 3, 4]pyrido[2, 1-*b*]quinazolin-5(7*H*)-one is suggested for the isolated methyl ester, the naturally occurring carboxy alkaloid being **1a**.

Work is continuing to confirm the proposed absolute configuration at C-13*b*⁷ and to search for a more efficient synthetic way to **1b**.



- 1 J.E. Saxton, in: The Alkaloids, vol.6 and references quoted therein. Ed. M.F. Grundon (Specialist Periodical Reports). The Chemical Society, London 1976.
- 2 B. Danieli, C. Farachi and G. Palmisano, *Phytochemistry* 15, 1095 (1976), and references quoted; B. Danieli and G. Palmisano, *J. heterocyclic Chem.* 14, 839 (1977).
- 3 M. Hesse, *Indolalkaloide*. Springer-Verlag, Berlin 1964.
- 4 R.T. Brown, C.L. Chapple and G.K. Lee, *J. chem. Soc. Chem. Commun.* 1972, 1007; R.T. Brown and A.A. Charalambides, *Tetrahedron Lett.* 17, 1649 (1974).
- 5 B. Danieli and G. Palmisano, *Heterocycles* 9, 803 (1978).
- 6 W. Klyne, R.J. Swan, N.J. Dastoor, A.A. Gorman and H. Schmid, *Helv. chim. Acta* 50, 115 (1966); W.F. Trager, C.M. Lee and A.H. Beckett, *Tetrahedron* 23, 365 (1967); C.M. Lee, W.F. Trager and A.H. Beckett, *Tetrahedron* 23, 375 (1967).
- 7 In this case it was not possible to establish the absolute configuration at C-13*b* by utilization of CD data (as for the yohimbine and heteroyohimbine alkaloids⁶) because the dipolar coupling between the electronic transitions of the indolic and anthranilic chromophores perturbed the Cotton effect sign in an unpredictable way.